

### Objective 13

WP: Determine the process for solving a multi-step problem using the units involved

1. Light travels  $9.00 \times 10^8$  m in 3 seconds in a vacuum. Which calculation would give the approximate time for light to travel 2.37 miles in a vacuum?

[A]  $\left(\frac{3 \text{ s}}{9.00 \times 10^8 \text{ m}}\right) \left(\frac{1000 \text{ m}}{\text{km}}\right) \left(\frac{\text{km}}{0.6214 \text{ mi}}\right) (2.37 \text{ mi})$

[B]  $\left(\frac{9.00 \times 10^8 \text{ m}}{3 \text{ s}}\right) \left(\frac{\text{km}}{1000 \text{ m}}\right) \left(\frac{0.6214 \text{ mi}}{\text{km}}\right) (2.37 \text{ mi})$

[C]  $\left(\frac{9.00 \times 10^8 \text{ m}}{3 \text{ s}}\right) \left(\frac{1000 \text{ m}}{\text{km}}\right) \left(\frac{0.6214 \text{ mi}}{\text{km}}\right) (2.37 \text{ mi})$

[D]  $\left(\frac{3 \text{ s}}{9.00 \times 10^8 \text{ m}}\right) \left(\frac{1000 \text{ m}}{\text{km}}\right) \left(\frac{0.6214 \text{ mi}}{\text{km}}\right) (2.37 \text{ mi})$

2. Light travels 558,140 miles in 4 seconds in water. How many seconds would it take for light to travel 13,000 m in water? Write the steps that would approximate this value. Use the fact that 1 m is approximately 39.37 inches.

3. A sample of zinc has a volume of  $4.67 \text{ cm}^3$  and a mass of 33.30 g. Which calculation would give the approximate weight in pounds of 1.72 cubic feet of zinc?

[A]  $\left(\frac{33.30 \text{ g}}{4.67 \text{ cm}^3}\right) \left(\frac{453.6 \text{ g}}{\text{lb}}\right) \left(\frac{\text{in}^3}{16.39 \text{ cm}^3}\right) \left(\frac{\text{ft}^3}{1728 \text{ in}^3}\right) (1.72 \text{ ft}^3)$

[B]  $\left(\frac{33.30 \text{ g}}{4.67 \text{ cm}^3}\right) \left(\frac{\text{lb}}{453.6 \text{ g}}\right) \left(\frac{16.39 \text{ cm}^3}{\text{in}^3}\right) \left(\frac{1728 \text{ in}^3}{\text{ft}^3}\right) (1.72 \text{ ft}^3)$

[C]  $\left(\frac{33.30 \text{ g}}{4.67 \text{ cm}^3}\right) \left(\frac{\text{lb}}{453.6 \text{ g}}\right) \left(\frac{16.39 \text{ cm}^3}{\text{in}^3}\right) \left(\frac{\text{ft}^3}{1728 \text{ in}^3}\right) (1.72 \text{ ft}^3)$

[D]  $\left(\frac{33.30 \text{ g}}{4.67 \text{ cm}^3}\right) \left(\frac{453.6 \text{ g}}{\text{lb}}\right) \left(\frac{16.39 \text{ cm}^3}{\text{in}^3}\right) \left(\frac{1728 \text{ in}^3}{\text{ft}^3}\right) (1.72 \text{ ft}^3)$

4. A sample of nickel has a volume of  $3.13 \text{ cm}^3$  and a mass of 27.86 g. Write the steps that would approximate the weight in pounds of 1.07 cubic feet of nickel. Use the fact that 1 pound is approximately 453.6 g and that 1 cubic inch is approximately  $16.39 \text{ cm}^3$ .

Light travels 279,070 miles in 2 seconds in water. Which calculation would give the approximate time for light to travel 15,000 m in water?

- [A]  $\left(\frac{279,070 \text{ mi}}{2 \text{ s}}\right)\left(\frac{\text{mi}}{5280 \text{ ft}}\right)\left(\frac{\text{ft}}{12 \text{ in.}}\right)\left(\frac{\text{m}}{39.37 \text{ in.}}\right)(15,000 \text{ m})$
- [B]  $\left(\frac{2 \text{ s}}{279,070 \text{ mi}}\right)\left(\frac{5280 \text{ ft}}{\text{mi}}\right)\left(\frac{\text{ft}}{12 \text{ in.}}\right)\left(\frac{39.37 \text{ in.}}{\text{m}}\right)(15,000 \text{ m})$
- [C]  $\left(\frac{279,070 \text{ mi}}{2 \text{ s}}\right)\left(\frac{5280 \text{ ft}}{\text{mi}}\right)\left(\frac{12 \text{ in.}}{\text{ft}}\right)\left(\frac{39.37 \text{ in.}}{\text{m}}\right)(15,000 \text{ m})$
- [D]  $\left(\frac{2 \text{ s}}{279,070 \text{ mi}}\right)\left(\frac{\text{mi}}{5280 \text{ ft}}\right)\left(\frac{\text{ft}}{12 \text{ in.}}\right)\left(\frac{39.37 \text{ in.}}{\text{m}}\right)(15,000 \text{ m})$

There is a 0.60 cubic foot sample of zinc. The sample weighs 267.00 pounds. Which calculation would give the approximate weight in grams of 13.46 cm<sup>3</sup> of zinc?

- [A]  $\left(\frac{267.00 \text{ lb}}{0.60 \text{ ft}^3}\right)\left(\frac{453.6 \text{ g}}{\text{lb}}\right)\left(\frac{\text{ft}^3}{1728 \text{ in}^3}\right)\left(\frac{\text{in}^3}{16.39 \text{ cm}^3}\right)(13.46 \text{ cm}^3)$
- [B]  $\left(\frac{267.00 \text{ lb}}{0.60 \text{ ft}^3}\right)\left(\frac{453.6 \text{ g}}{\text{lb}}\right)\left(\frac{1728 \text{ in}^3}{\text{ft}^3}\right)\left(\frac{16.39 \text{ cm}^3}{\text{in}^3}\right)(13.46 \text{ cm}^3)$
- [C]  $\left(\frac{267.00 \text{ lb}}{0.60 \text{ ft}^3}\right)\left(\frac{\text{lb}}{453.6 \text{ g}}\right)\left(\frac{1728 \text{ in}^3}{\text{ft}^3}\right)\left(\frac{\text{in}^3}{16.39 \text{ cm}^3}\right)(13.46 \text{ cm}^3)$
- [D]  $\left(\frac{267.00 \text{ lb}}{0.60 \text{ ft}^3}\right)\left(\frac{\text{lb}}{453.6 \text{ g}}\right)\left(\frac{\text{ft}^3}{1728 \text{ in}^3}\right)\left(\frac{\text{in}^3}{16.39 \text{ cm}^3}\right)(13.46 \text{ cm}^3)$

A motor makes 8775 complete rotations in 3.9 minutes. Which calculation would give the number of degrees the motor rotates in 19.8 seconds?

- [A]  $\left(\frac{8775 \text{ rotations}}{3.9 \text{ min}}\right)\left(\frac{\text{min}}{60 \text{ s}}\right)\left(\frac{\text{rotation}}{360 \text{ degrees}}\right)(19.8 \text{ s})$
- [B]  $\left(\frac{8775 \text{ rotations}}{3.9 \text{ min}}\right)\left(\frac{60 \text{ s}}{\text{min}}\right)\left(\frac{360 \text{ degrees}}{\text{rotation}}\right)(19.8 \text{ s})$
- [C]  $\left(\frac{8775 \text{ rotations}}{3.9 \text{ min}}\right)\left(\frac{\text{min}}{60 \text{ s}}\right)\left(\frac{360 \text{ degrees}}{\text{rotation}}\right)(19.8 \text{ s})$
- [D]  $\left(\frac{8775 \text{ rotations}}{3.9 \text{ min}}\right)\left(\frac{60 \text{ s}}{\text{min}}\right)\left(\frac{\text{rotation}}{360 \text{ degrees}}\right)(19.8 \text{ s})$